

CHEM 14D Midterm 2
November 19, 2018

Full Name on Every Page* *Write Dark* *Only front pages will be scanned

Name (Last) _____ (First) _____ Student ID _____
Lecture _____

Question	1	2	3	4	5	FULL Name on every page	Total
Points	12	20	25	20	21	2	100

- 1. Write your full name on every page.**
- 2. Answers written on the back of the exam will not be graded.**
- 3. This is a closed book exam. The use of notes, cell phones, calculators, or other devices will not be allowed during exam.**
- 4. Where applicable, answers without a clear indication of stereochemistry will not be given any credit.**
- 5. You may use model sets brought in a clear ziplock bag.**
- 6. For full credit show your work, partial credit will be awarded.**
- 7. A periodic table will be provided to you.**
- 8. Show your student ID when turning in your exam.**

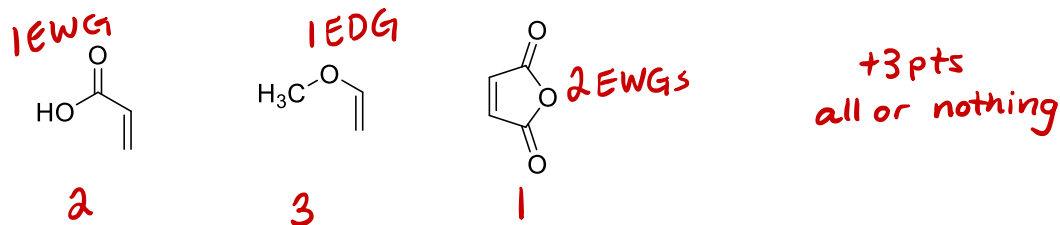
1.1. (3 points) Circle the option that best completes the sentence.

Reacting an alkyne with 2 equivalents of HCl gives a (vicinal **geminal**) dichloride. *1pt.*

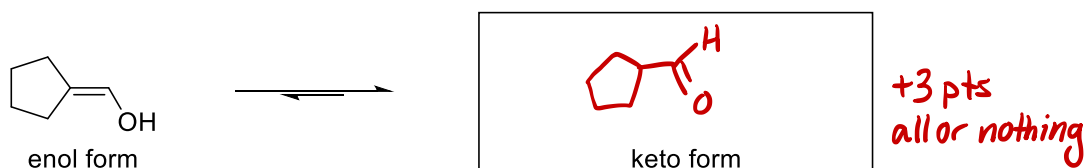
Alkenes typically undergo **addition** / **elimination**) reactions with electrophiles. *1pt.*

Markovnikov's rule can be used to describe (stereo- / **regio-**) selectivity. *1pt.*

1.2. (3 points) Rank the following dienophiles in order of reactivity in a Diels-Alder reaction. (1 = most reactive)



1.3. (3 points) Show the product of tautomerization of the following enol to its keto form.



1.4. (3 points) Circle true or false for the following statements:

Reacting OsO₄ with an alkene adds two -OH groups with anti addition.

True or **False** *(syn)*

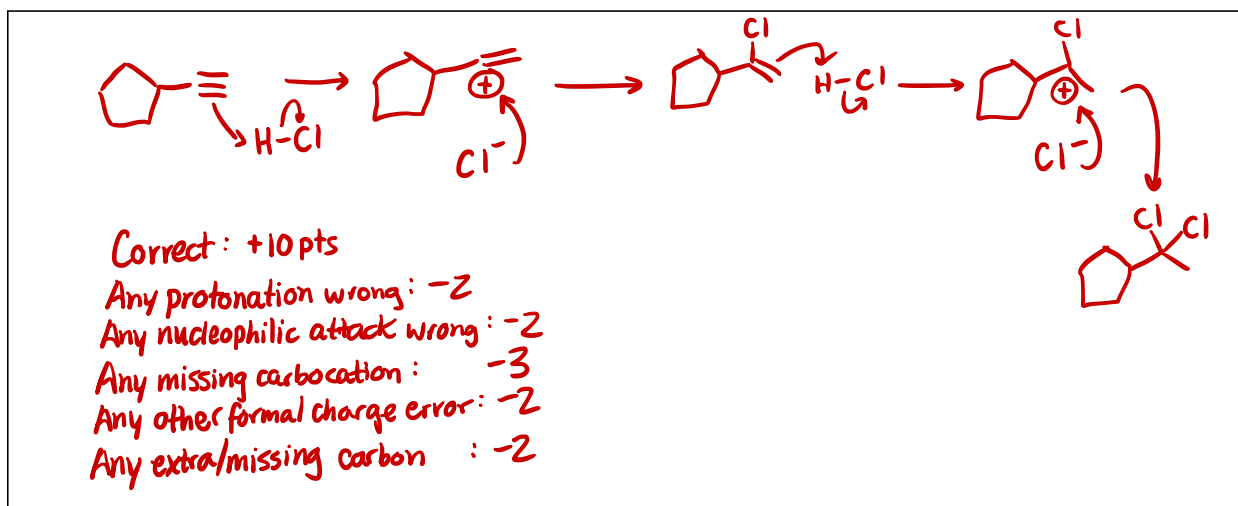
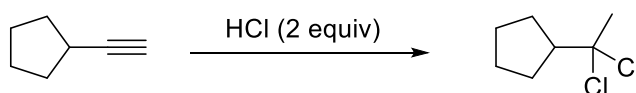
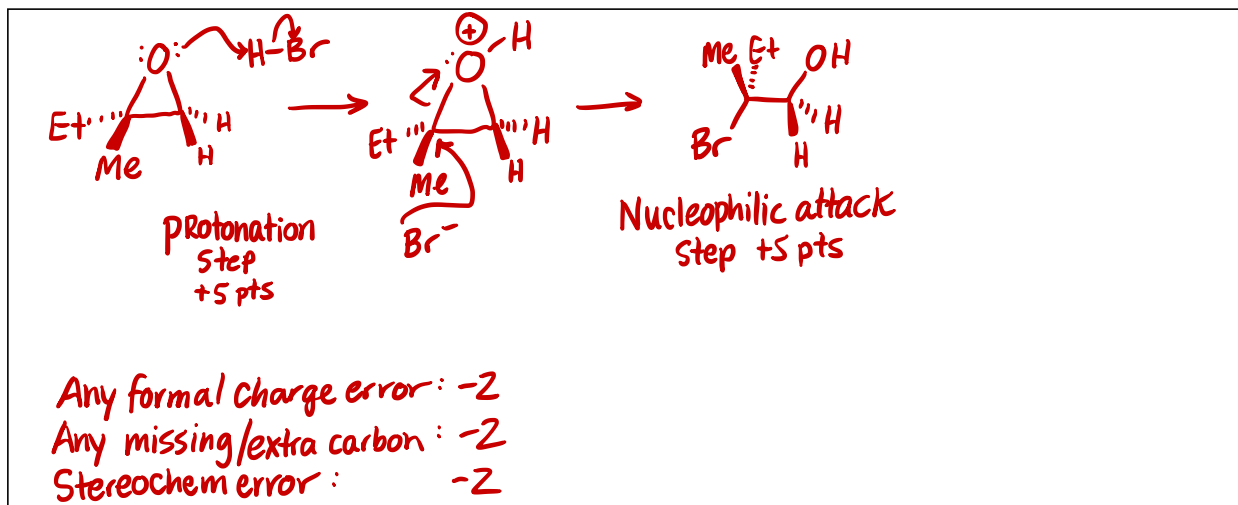
The pK_a of an alkyne is 25.

True or False

Epoxides are attacked by Me-Li at the most-substituted carbon.

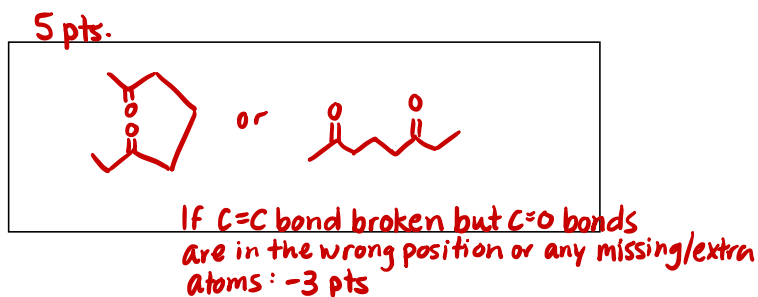
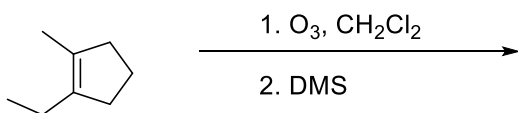
less
True or **False**

2.1. (20 points) For each part, provide a detailed arrow-pushing reaction mechanism for the reaction shown.

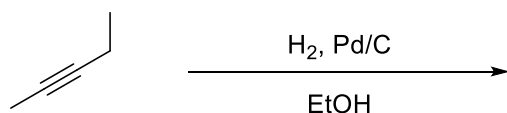


3.1. (25 points) Predict the major organic product(s) derived from the substrates shown to the left of the arrow in the following reactions. **If multiple stereoisomers would form, just draw one.**

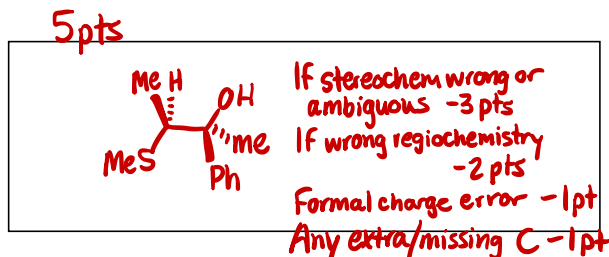
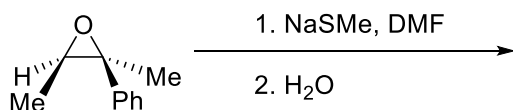
(a)



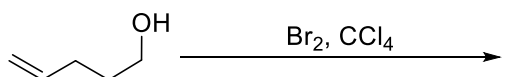
(b)



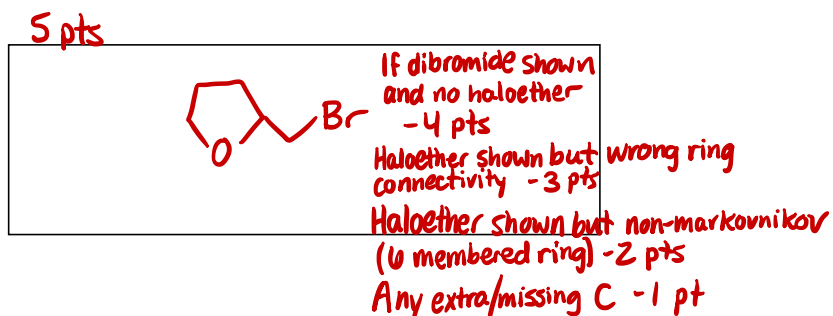
(c)



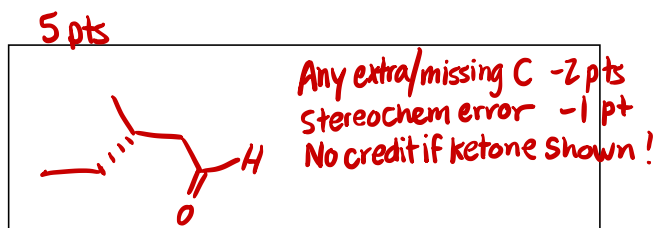
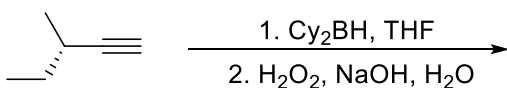
(d)



Haloether formation
See Practice MT 2.1b

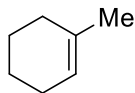


(e)



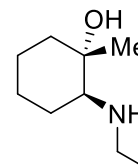
4.1. (20 points) Predict the missing reagent(s) to accomplish the following transformations.

(a)



4 pts

1. mCPBA 2 pts
 2. NaNH₂ (E+NH₂ also ok) 2 pts
 3. H₂O ← No workup after NaNH₂? -1 pt.



From Syllabus

+ enantiomer

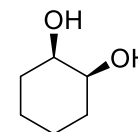
Did not use separate steps but otherwise correct? -2 pts

(b)



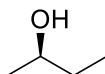
4 pts

1. OsO₄, THF
 2. H₂S



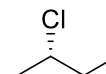
If no H₂S -2 pts
 Combined step 1+2 -1 pt

(c)



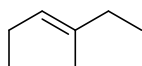
4 pts

SOCl₂, Et₃N



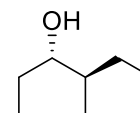
Missing Et₃N -1 pt
 Used H-Cl -2 pts
 (wrong stereochem)

(d)



4 pts

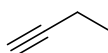
1. BH₃, THF +2 pts
 2. H₂O₂, NaOH, H₂O +2 pts



+ enantiomer

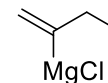
No separate steps -1

(e)



optional

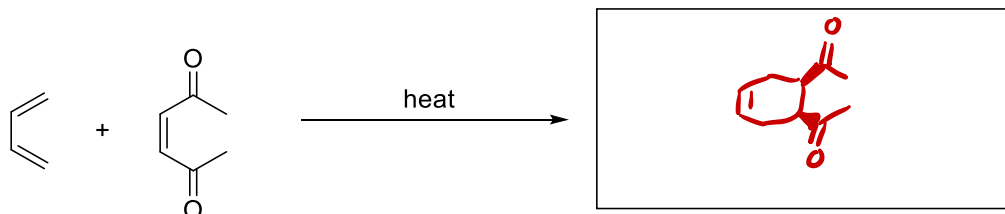
1. H-Cl (1 equiv) 2 pts
 2. Mg, Et₂O 2 pts



MgCl

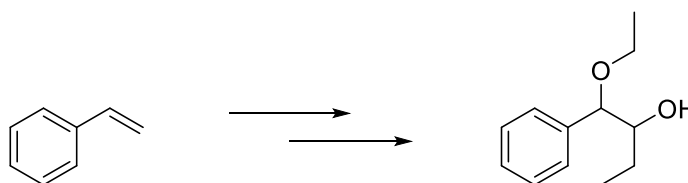
If said "2 equiv" -1 pt
 No separate steps -1 pt
 Protic solvent w/ Mg -1 pt

5.1. (6 points) Draw the product of the following Diels-Alder reaction (showing relative stereochemistry).

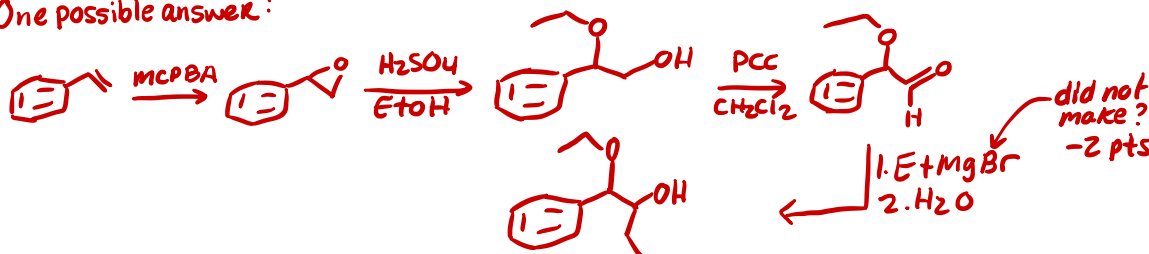


*Incorrect stereochem
or no stereochem
- 2 pts*

5.2. (15 points) Propose syntheses for the molecule shown from the indicated starting material. You may also use fragments of 2-carbons or less. These fragments may only contain combinations of the following atoms: C, H, O, Br, I, Cl (with single or double bonds). Please be sure to include key reagents and solvents in your solutions, consistent with what you have learned in this course. If you must use an organometallic, show how it's made. It is not necessary to show mechanisms or retrosynthetic analyses in your solutions. Hint: There is often more than one correct answer! **Don't worry about stereochemistry.**



One possible answer:



Mostly correct, but minor issue (e.g. 1 missing/wrong reagent but intention is clear and otherwise correct.) -3 pts

Partial Plan to product (e.g. formation of epoxide, Williamson Ether Synth., or haloether synthesis) but overall did not work. -10 pts.

Any use of OsO_4 and then an attempt to react only 1 -OH group -10 pts

At some point, if an aldehyde was made and attacked w/ Et-MgBr or Et-Li +3 pts

Used illegal reagent (more than 3 C's) but overall correct -13 pts.

Any missing workup -2 pts Any formal charge error -1 pt.

Any 1 missing/extra carbon that does not have major impact on plan -1 pt.