

Chem. 20A

Quiz #1

Professor Baugh

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Student

Discus:

Directic. **k must** be shown in order to get credit. The correct SI units must be used. Write on the back of the page if needed. Good luck!

Question #1: Coulombs Law the Electric Potential Energy

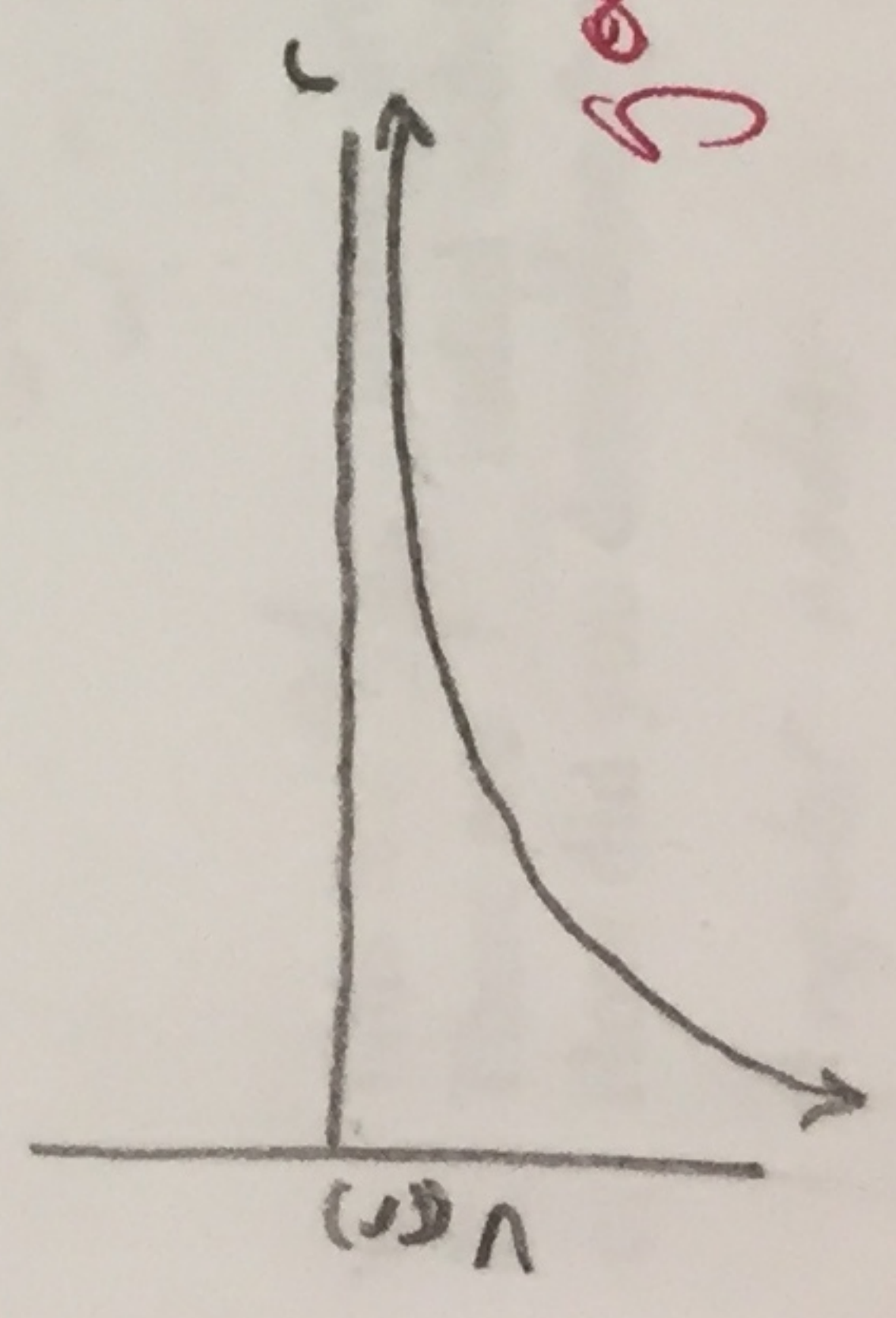
A Helium nucleus and an electron are located 5.0×10^{-7} m apart

- a. What is the electric potential energy between the two particles? Is the PE attractive or repulsive? (20 pts.) *It is attractive. The potential is -9.2×10^{-22} J.*
- b. How far apart are the particles when the energy between the particles is 7.0eV (10 pts) *They are 4.1×10^{-10} m apart.*
- c. **Sketch** a plot of the electric potential energy vs. separation distance and explain the asymptotic behavior of this function (i.e. what happens to the potential as $r \rightarrow 0$ and as $r \rightarrow \infty$. (20 pts)

$$\lambda = \frac{h}{mv} = \frac{6.626 \times 10^{-34} \text{ J}\cdot\text{s}}{(9.109 \times 10^{-31} \text{ kg})(2 \times 10^6 \text{ m/s})} = 3.6 \times 10^{-10} \text{ m}$$

Question #2: Basics of Quantum Mechanics

- a. What is the De Broglie wavelength of an electron with velocity 2×10^6 m/s? (15 pts.) *3.6 x 10⁻¹⁰ m*
- b. Regarding the Heisenberg uncertainty principle, if the standard deviation (or "spread") of the values in the position of an electron is 3.5×10^{-4} m, what is the minimum uncertainty in the speed of the particle? What about for a basketball that weighs .56 kg? (20 pts.) Interpret your result from part b (15 pts) *5*



As $r \rightarrow 0$, the particles become exponentially attracted to each other. All of the energy in the system is potential energy. When $r \rightarrow \infty$, the particles become disassociated, and are not "aware" of the other. All energy in this case is kinetic. \rightarrow so the PE $\rightarrow 0$

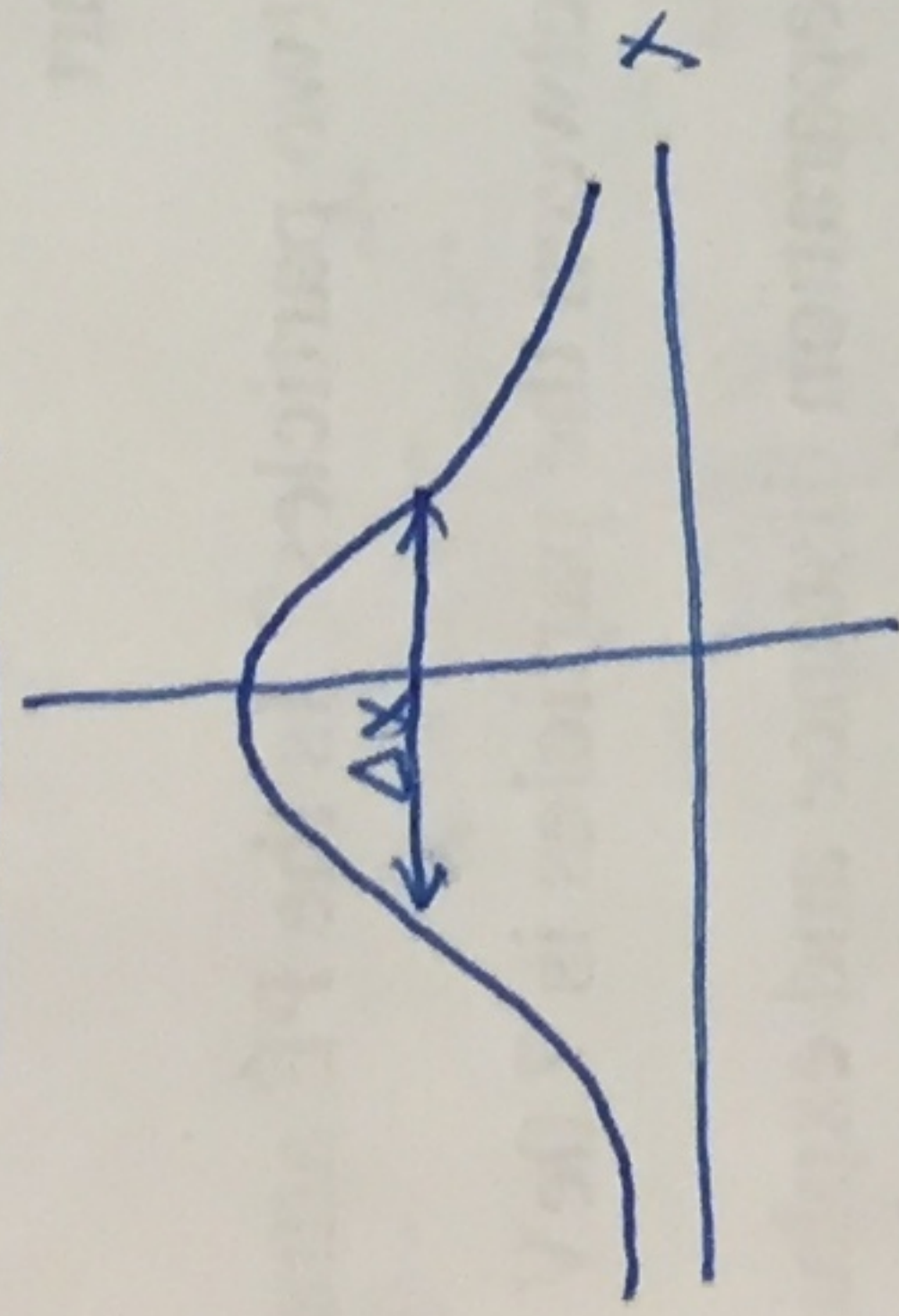
(2b) $\Delta v_e \geq 1.4 \text{ m/s}$
 $\Delta v_b \geq 2.7 \times 10^{-31} \text{ m/s}$

(2c) (on back) *5*

79
84

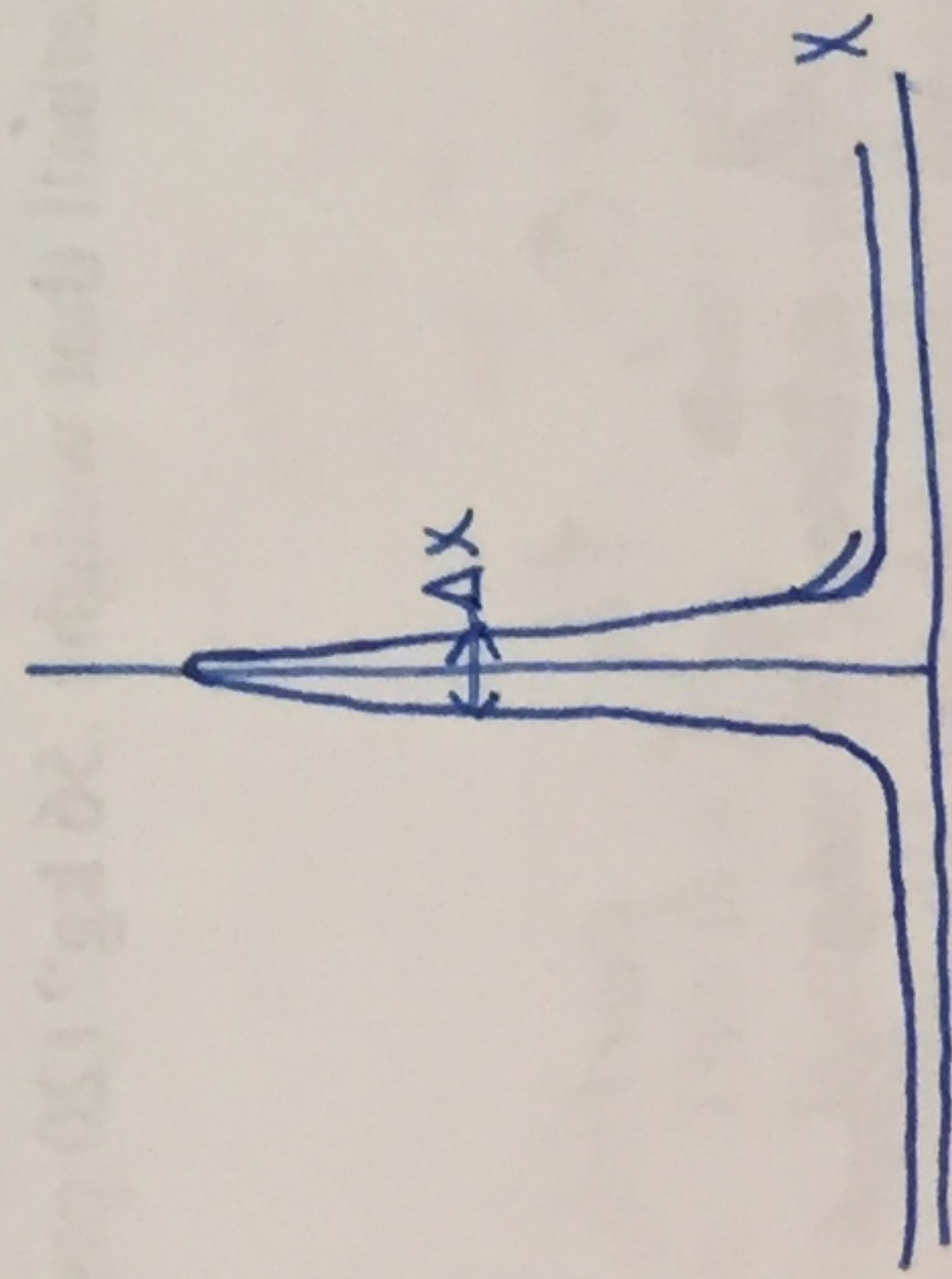
2c) For an electron which has a very small mass, the given change in position is a lot, proportional to its size. Thus, the change in velocity can be much greater than that of a basket ball. For a basket ball, such a small change in position compared to its much greater mass and size results in a smaller change in velocity.

Probability



electron

(✓ Right answer)
explanation)



basket ball

★ ↑ accuracy of position, ↓ accuracy of momentum, & vice versa,
shining light on a particle (electron) changes momentum, however for large